

COURSE OUTLINE

1. GENERAL INFORMATION

SCHOOL	MARITIME AND INDUSTRIAL STUDIES		
DEPARTMENT	INDUSTRIAL MANAGEMENT AND TECHNOLOGY		
LEVEL OF STUDY	UNDERGRADUATE		
COURSE UNIT CODE	TEΦΥΣ02-2	SEMESTER OF STUDY	2 nd
COURSE TITLE	PROCESSES I		
INDEPENDENT TEACHING ACTIVITIES <i>in case in which credits are awarded for separate components/parts of the course, e.g. in lectures, laboratory exercises, etc. If credits are awarded for the whole of the course, give the weekly teaching hours and the total credits</i>		WEEKLY TEACHING HOURS	CREDITS
Lecture, Laboratory Exercises and Project			5.5
<i>Add rows if necessary. The organization of teaching and the teaching methods used are described in detail at section 4.</i>			
COURSE TYPE <i>general background, special background, specialized general knowledge, skills development</i>	Special background		
PREREQUISITE COURSES:	None		
LANGUAGE OF INSTRUCTION and EXAMINATION/ASSESSMENT:	Greek		
THE COURSE IS OFFERED TO ERASMUS STUDENTS	No		
COURSE WEBSITE (URL)	https://eclass.unipi.gr/courses/BDT233/		

2. LEARNING OUTCOMES

<p>LEARNING OUTCOMES</p> <p><i>The course learning outcomes, specific knowledge, skills and competences of an appropriate (certain) level, which students will acquire upon successful completion of the course, are described in detail. It is necessary to consult:</i></p> <p>APPENDIX A</p> <ul style="list-style-type: none"> • <i>Description of the level of learning outcomes for each qualifications' cycle, according to the European Higher Education Area's Qualification Framework.</i> • <i>Descriptors for Levels 6, 7 & 8 of the European Qualifications Framework for Lifelong Learning and APPENDIX B</i> • <i>Guidelines for writing Learning Outcomes</i>
<p>The course aims to familiarize students with:</p> <ul style="list-style-type: none"> • The basic chemical processes and flow diagrams of representative manufacturing processes • Scaling up from bench to pilot and industrial scale • Scale economies and externalities in industrial combos • Material and energy balances at steady and non-steady state conditions • The chemical processes in environmental protection • Material saving with recycling • Energy saving with thermochemical and electrochemical systems • The dynamic physicochemical systems, robustness, sensitivity, risk, economic and technical optimization • The correlation between manufacturing parameters and product quality • The optimization of production capacity • Laboratory and simulation applications

Upon successful completion of the course, the students will be able to:

- Solve mass balance problems at steady and non-steady state conditions
- Present basic processes in flow charts
- Solve chemical equilibrium problems
- Solve capacity optimization problems
- Problem solving in chemical production and environmental protection

General Competences

Taking into consideration the general competences that students/graduates must acquire (as those are described in the Diploma Supplement and are mentioned below), at which of the following does the course attendance aims

Search for, analysis and synthesis of data and information, by the use of technologies that are necessary according the case

Adapting to new situations

Decision-making

Independent work

Team work

Working in an international environment

Working in an interdisciplinary environment

Introduction of innovative research

Project planning and management

Respect for difference and multiculturalism

Environmental awareness

Social, professional and ethical responsibility and sensitivity to gender issues

Critical consciousness, criticism and self-criticism

Development of free, creative and inductive thinking

- Search for, analysis and synthesis of data and information, by the use of technologies that are necessary according the case
- Adapting to new situations
- Decision-making
- Independent work
- Team work
- Respect for difference and multiculturalism
- Environmental awareness
- Social, professional and ethical responsibility and sensitivity to gender issues
- Critical consciousness, criticism and self-criticism

2. COURSE CONTENT

The course includes lectures and laboratory.

Week	Topics
1 st	Basic chemical processes and flow diagrams of representative manufacturing processes.
2 nd	Scaling up from bench to pilot and industrial scale.
3 rd	Scale economies and externalities in industrial combos.
4 th	Material and energy balances at steady and non-steady state conditions.
5 th	Chemical processes in environmental protection. Material saving with recycling.
6 th	Energy saving with thermochemical and electrochemical systems.
7 th	Dynamic physicochemical systems, robustness, sensitivity, risk, economic and technical optimization.
8 th	Correlation between manufacturing parameters and product quality.
9 th	Optimization of production capacity: methodology and applications.
10 th	Batch reactor models and simulation.
11 th	CFSTR models and simulation.
12 th	PFR models and simulation.
13 th	Review exercises

Laboratory: Simulation and optimization of processes using physical simulators in the Laboratory of Simulation of Industrial Processes

Week			Laboratory Syllabus
1 st	5 th	9 th	Economic and technical optimization of CFST reactors in series
2 nd	6 th	10 th	Adsorption column optimization. Aluminium anodizing optimization
3 rd	7 th	11 th	Optimization of batch reactor for biomass processing
4 th	8 th	12 th	PFR modeling and implementation in wastewater treatment.
13 th			Rescheduled lab practice

Students also attend a laboratory training program in the Laboratory of Simulation of Industrial Processes in order to develop an intuitive and hands-on understanding of the concepts presented in the lectures, such as modeling of industrial processes, data analysis and decision making based on techno-economic criteria. The software used is MS EXCEL or equivalent (Open Office, etc.) as well as in house software. Students are trained in workshops with a rotation system. The workshop program is posted on the course website and eclass at the beginning of the semester.

In addition, articles, audiovisual lecture material, web addresses, useful information and exercises are posted at eclass.

3. TEACHING METHODS - ASSESSMENT

<p style="text-align: center;">TEACHING MODE</p> <p><i>Face-to-face, in-class lecturing, on distance teaching and distance learning etc.</i></p>	In-class lecturing / Laboratory teaching																	
<p style="text-align: center;">USE OF INFORMATION AND COMMUNICATION TECHNOLOGY</p> <p><i>Use of ICT in Teaching, Laboratory Education, Communication with students</i></p>	<p>Teaching: Lectures with audiovisual media, support of the learning process through the eclass platform</p> <p>Laboratory Education: Use of open access and in-house software for laboratory exercises</p> <p>Communication with students: face-to-face at office hours, email, eclass</p>																	
<p style="text-align: center;">COURSE DESIGN</p> <p><i>Description of teaching techniques, practices and methods:</i></p> <p><i>Lectures, seminars, laboratory practice, fieldwork, study and analysis of bibliography, tutorials, clinical practice, Art Workshop, Interactive teaching, Educational visits, project, Essay writing, Artistic creativity, etc.</i></p> <p><i>The study hours for each learning activity as well as the hours of non- directed study are given according to the principles of the ECTS</i></p>	<table border="1" style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th style="text-align: center;"><i>Activity / Method</i></th> <th style="text-align: center;"><i>Semester Workload</i></th> </tr> </thead> <tbody> <tr> <td>Lectures</td> <td style="text-align: center;">52</td> </tr> <tr> <td>Laboratory exercises</td> <td style="text-align: center;">8</td> </tr> <tr> <td>Project</td> <td style="text-align: center;">30</td> </tr> <tr> <td>Self-study of lecture and lab material</td> <td style="text-align: center;">45</td> </tr> <tr> <td>Counselling</td> <td style="text-align: center;">0.5</td> </tr> <tr> <td>Exams (written)</td> <td style="text-align: center;">2</td> </tr> <tr> <td>Course Total</td> <td style="text-align: center;">137.5</td> </tr> </tbody> </table>		<i>Activity / Method</i>	<i>Semester Workload</i>	Lectures	52	Laboratory exercises	8	Project	30	Self-study of lecture and lab material	45	Counselling	0.5	Exams (written)	2	Course Total	137.5
	<i>Activity / Method</i>	<i>Semester Workload</i>																
	Lectures	52																
	Laboratory exercises	8																
	Project	30																
	Self-study of lecture and lab material	45																
	Counselling	0.5																
	Exams (written)	2																
Course Total	137.5																	
<p>STUDENT PERFORMANCE EVALUATION/ASSESSMENT METHODS</p> <p><i>Detailed description of the evaluation procedures:</i></p> <p><i>Language of evaluation, assessment methods, formative or summative (conclusive), multiple choice questionnaires, short- answer questions, open-ended questions, problem solving, written work, Essay/report, oral exam, public presentation, laboratory work, art interpretation, other.....etc</i></p> <p><i>Evaluation criteria are specifically defined and given as well as if and where they are reported and accessible to students.</i></p>																		

institution) is performed according to the relevant procedure decided by the Department Assembly.

Notification of the Assessment Criteria: The evaluation criteria are made known during the first lecture and are clearly stated on the course website and e-class. The answers to the exam questions are posted at eclass after the exam date. Students have the opportunity to discuss their exam paper with the course instructor (at the posted office hours) after the announcement of the course grades.

4. SUGGESTED BIBLIOGRAPHY

-Suggested Bibliography :

- Book [9722]: Chemical Industry Design [in Greek], D.S. Marinos- Kouris, Z.B. Maroulis
- Book [18549018]: Elements of Chemical Processes [in Greek], K.A. Matis, P. Mavros, K.S. Triantafillidis

-Scientific Journals: not applicable

-Lecture Notes

-Laboratory Workbook